

UNIVERSITAS NEGERI YOGYAKARTA

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Bachelor of Education in Science

MODULE HANDBOOK

Module name:	Modern Physics					
Module level, if applicable:	Undergraduate					
Code:	IPA6241					
Sub-heading, if applicable:	-					
Classes, if applicable:	-					
Semester:	6 th					
Module coordinator:	Dr. Dadan Rosana, M.Si					
Lecturer(s):	Dr. Dadan Rosana, M.Si					
Language:	Bahasa Indonesia					
Classification within the curriculum:	Compulsory Course					
Teaching format / class hours per week during the semester:	100 minutes lectures and 120 minutes structured activities per week.					
	Total workload is 90,67 hours per semester which consists of					
Workload:	100 minutes lectures, 120 minutes structured activities, and					
	120 minutes individual study per week for 16 weeks.					
Credit points:	2 SKS (3 ETCS)					
Prerequisites course(s):	-					
Targeted learning outcomes:	After careful study of this chapter you should be able to do the following: CO1. Understand the concept of Reference point. CO2. define the Special and general theories of relativity. CO3. make Galileo conversions CO4. be knowledgeable about the propositions of relativity CO5. To establish a relationship between mass and energy CO6. Understand the Doppler Effect CO7. comprehend the particle properties of waves. CO8. Understand the wave properties of particles CO9. know the uncertainty principle. CO10. know the Atomic structure and development					

	CO11. Describe the series of spectra CO12. know Frank-Hertz experiment and know the basic operating principle of the laser.						
Content:	The Modern Physics course is a compulsory subject for students elective course in Science Education. In this course, facilitated by lecturers, students will study; 1 Understand the concept of Reference point. 2 define the Special and general theories of relativity. 3 make Galileo conversions 4. be knowledgeable about the propositions of relativity 4.1. be knowledgeable about the concept of time dilation. 4.2. be knowledgeable about the concept of length contraction. 4.3. be able to Lorentz transformations. 4.4. Know the concept of relativistic mass 5 To establish a relationship between mass and energy 6 Understand the Doppler Effect 7 comprehend the particle properties of waves. 7.1. explain the Photoelectric effect, Compton effect and Couples. 7.2. know the quantum theory of light 7.3. Recognize the X-ray diffraction and Brag's law. 8 Understand the wave properties of particles 8.1. Know the concept of the wave function 8.2. be knowledgeable about Particle diffraction (Davisson -Germer experiment). 8.3. know the behavior of particles in the box 9 know the uncertainty principle. 10 know the Atomic structure and development. 10.2. Know the Thomson and Rutherford atomic model. 10.3. Have information about the Bohr atom model. 10.5. Will learn the core of the movement 11.1. Describe the series of spectra 12.1. know Frank-Hertz experiment 12.2. know the basic operating principle of the laser						
Study / exam achievements:	Attitudeassessmentiscarriedoutateachmeetingbyobservationand/orself-assessmenttechniquesusingtheassumptionthatbasicallyeverystudenthasa goodattitude.Thestudentisgiven a value ofverygood or notgoodattitude.Thestudentisgiven a value ofverygood or notgoodattitudeiftheyshowitsignificantlycomparedtootherstudentsingeneral.Theresult ofattitudeassessmentisnotacomponentofthefinalgrades,butasoneoftheteasthaveagoodattitude.Thefinalgrades,butasfond <t< td=""></t<>						

		CO9 CO10 CO11 And CO12			1000/			
Forms of modia:	Poord LCD Projector Lepton/Computer							
	Board, LCD Projector, Laptop/Computer							
Literature:	 K. Krane, Modern Physics 3rd ed, Wiley, 2012. R. Harris, Modern Physics 2nd ed., Pearson, 2008. J. Bernstein, P. M. Fishbane, and S. Gasiorowicz, Modern Physics, Prentice Hall, 2000 Serway, Moses, and Moyer (2005). Modern Physics. Third Edition, Brooks/Cole. 							

PLO and CO mapping

	PLO											
	Attitude			Knowledge				Spesific SKill				
	PLO1	PLO2	PLO3	PLO1	PLO2	PLO3	PLO4	PLO1	PLO2	PLO3	PLO4	PLO5
CO1						✓	✓					
CO2				\checkmark		\checkmark						
CO3					✓	\checkmark						
CO4					✓		✓					
CO5				✓			✓					
CO6					✓	✓						
C07				✓	✓		✓					
CO8				✓		✓	✓					
CO9						✓						
CO10				✓	✓		✓					
CO11				\checkmark	✓		✓					
CO12				\checkmark		✓	\checkmark					